

# QUANTIFYING SEDIMENT METHANOGENESIS AND CARBON DIOXIDE EMISSIONS FROM MACROPHYTE PLANT COMMUNITIES IN A TIDAL HUDSON RIVER WETLAND

CURT MCCONNELL

*Ithaca College, Ithaca, NY 14850 USA*

MENTOR SCIENTIST: DR. STUART E.G. FINDLAY

*Cary Institute of Ecosystem Studies, Millbrook, NY 12545 USA*

*Abstract.* Wetlands are the largest natural source of global methane emissions, most of which are generated from methane-producing macrophyte beds. Methane, a potent greenhouse gas, is released from anaerobic methanogenic archaea that break down organic matter. The amount of organic carbon available to methanogens is dependent on the carbon content of the sediments associated with the plant community; therefore, it is important to identify macrophyte populations that have the greatest influence on methanogenesis. Depending on the combination of other factors including dissolved oxygen, dissolved organic carbon, and nitrate in the sediment, wetlands can also be significant sources of CO<sub>2</sub>. In this study CH<sub>4</sub> and CO<sub>2</sub> emissions were analyzed from four different macrophyte sediment beds in the tidal Hudson River's Tivoli Bay wetlands. *Vallisneria americana*, *Typha angustifolia*, *Trapa natans*, and *Phragmites australis* were studied. Methane emissions ranged from 3.8±4.4 to 11.0±9.7 ppm/112cm<sup>2</sup>/min in *P. australis*, *T. angustifolia*, and *T. natans*. *V. americana* had the lowest methane emissions at ~0.32ppm/112cm<sup>2</sup>/min. Mean CO<sub>2</sub> release from the sediment cores was the greatest from *P. australis* with 15.5±8.3 ppm/112cm<sup>2</sup>/min, 10.5±4.5 ppm/112cm<sup>2</sup>/min for *T. natans*, 10.5±2.9 ppm/112cm<sup>2</sup>/min for *T. angustifolia*, and 1.2±0.9 ppm/112cm<sup>2</sup>/min for *V. americana*. The differences in methane emissions were not significant between the four types, but there is sufficient evidence to conclude that if the goal of conservation or construction of artificial wetlands is reducing CH<sub>4</sub> and CO<sub>2</sub> emissions, *V. americana* would be the most ideal species to conserve or plant (conditions willing). Future studies should incorporate other sediment methanogenesis parameters, such as ethanol content and microbial communities, to gain a comprehensive understanding of this complex process. All of these factors need to be kept into consideration as climate change and subsequent changes in hydrology could dramatically alter the current trends in wetland greenhouse gas exchange.

## INTRODUCTION

Apart from humans who emit 54-72% of the world's methane (CH<sub>4</sub>) gas (Sharifi et al. 2013), it is estimated that wetlands are responsible for 10-40% of the global CH<sub>4</sub> flux, while only making up 4-6% of the earth's land area (Chowdhury and Dick 2013; Grünfeld and Brix 1999; Mitsch and Gosselink 2000; Tian, et al. 2011). With a radiative forcing 25 times greater than that of carbon dioxide (CO<sub>2</sub>) (Bridgman et al. 2013; Chowdhury and Dick 2013), CH<sub>4</sub> is a significant contributor to the greenhouse effect. Although a less potent greenhouse gas, wetlands emit greater amounts of CO<sub>2</sub> than CH<sub>4</sub> (Clair et al. 2002). Because of the greenhouse gas effects from CH<sub>4</sub> and CO<sub>2</sub> it is important to identify the major contributors to wetland greenhouse gas production and to quantify emissions.

Depending on the type and density, macrophyte species are significant contributors to CH<sub>4</sub> and CO<sub>2</sub> production, albeit indirectly. Decaying macrophytes supply organic carbon to bacteria and archaea, which in turn, either produce CH<sub>4</sub> or CO<sub>2</sub> during the biodegradation of plant detritus. When terminal electron acceptors used in the ATP production pathway such as oxygen or nitrate are present in the sediment, CO<sub>2</sub> dominates as the byproduct of microbial biodegradation. CH<sub>4</sub>, on the other hand, is produced as a fermentative byproduct through a process called methanogenesis (Boon et al. 1995). Methanogens use small carbon molecules as the terminal electron acceptors in the electron transport chain, and the carbon

molecules are converted into CH<sub>4</sub> (Chowdhury and Dick 2013; Sutton-Grier and Megonigal 2011; Tian et al. 2011) Methanogenesis occurs anaerobically so it is ubiquitous in the hydric sediments of wetlands. With 269,252 acres of wetlands in the Hudson River Valley, CH<sub>4</sub> and CO<sub>2</sub> release is predictably substantial (Grigg 2010). In Hudson River wetlands, sediment methanogenesis and CO<sub>2</sub> emission rates were compared between four different macrophyte populations to better understand how aquatic vegetation affects the local greenhouse gas budget.

It is expected that macrophyte populations would differ in CH<sub>4</sub> and CO<sub>2</sub> emissions due to varying sediment characteristics; however, this study focused on the influences of dissolved oxygen (DO), pore-water nitrate, pore-water dissolved organic carbon (DOC), and total organic carbon (TOC) content of the macrophyte sediment beds. When DO is abundant, aerobic decomposition will predominate, preventing the anaerobic methanogenesis process from occurring. CO<sub>2</sub> will then be the dominant byproduct from the aerobic breakdown of organic matter. DO varies spatially as some macrophyte beds may facilitate the presence of oxygen more so than others, leading to differing levels of CH<sub>4</sub> and CO<sub>2</sub> production (Kao-Kniffin et al. 2010). Measuring nitrate was important because it is the most efficient terminal electron acceptor in the anaerobic decomposition pathway of organic matter (Sutton-Grier and Megonigal 2011). If nitrate is present in the sediment, facultative or obligate anaerobic microbes will select nitrate instead of small carbon molecules as the terminal electron acceptors in the fermentation process, inhibiting CH<sub>4</sub> production and leading to CO<sub>2</sub> production.

High DOC in the pore water as well as high TOC in the sediment influences methanogenesis and CO<sub>2</sub> production by providing the fuel source for microbes. Levels of TOC and DOC may differ based on the overall population density of macrophyte species. The DOC in the pore water also depends on the sediment type and how much easily degradable organic matter is present (Jespersen et al. 1998; Miyajima et al. 1997). The amount of readily available DOC is estimated by the decomposability of plant species (Tian et al. 2011). Those that decay the fastest produce the most organic carbon quickly. *T. angustifolia* and *P. australis* have much lower decomposition rates than that of *V. americana* and *T. natans*. *T. angustifolia* has a decomposition rate of about 0.0023d<sup>-1</sup>. The decay rate of *P. australis* is 0.0039d<sup>-1</sup> (Chimney and Pietro 2006) *T. natans* is 0.014d<sup>-1</sup> (Findlay, et al. 1990) and *V. americana* has the greatest decay rate of 0.0931d<sup>-1</sup> (Chimney and Pietro 2006). Since *T. angustifolia* has the lowest decomposition rate, it has the greatest amount of lignin and fibrous cellulose/hemicellulose. These structural fibers are much less biodegradable (Miyajima et al. 1997) therefore *T. angustifolia* will most likely produce less DOC than the other species.

Producing the maximum amount of DOC is important, particularly for methanogens, because the terminal electron acceptors ultimately used by the methanogenic archaea are small DOCs leached from macrophyte plant detritus (Boon and Mitchell 1995; Sutton-Grier and Megonigal 2011; Tian et al. 2011). The most readily accessible forms of DOC are simple organic molecules such as methanol, trimethylamine, acetate, and carbon dioxide/hydrogen (Chowdhury and Dick 2013; Tian et al. 2011). However, using organic carbon as the terminal electron acceptor is not the most efficient way of producing energy. In aerobic conditions, oxygen is preferentially used as the terminal electron acceptor by facultative aerobic microbes due to its efficiency as an electron acceptor. When oxygen is depleted the next most efficient electron acceptors are nitrate, sulfate, and ferric iron, respectively (Sutton-Grier and Megonigal 2011). CO<sub>2</sub> and hydrogen are produced as byproducts using these alternate electron acceptors. When the levels of these anaerobic alternate electron acceptors deplete in the sediment, CO<sub>2</sub> and other carbon-based molecules predominate as the terminal electron acceptors, creating an ideal environment for methanogens (Sutton-Grier and Megonigal 2011; Tian et al. 2011).

Which organic carbon molecules are used is dependent on the particular family of archaea. The Methanosaetaceae and Methanosarcina families are acetoclastic while the other methanogenic archaea are hydrogenotrophic (Kao-Kniffin et al. 2010; Tian et al. 2011). Independent of the family of methanogen,

the amount of CH<sub>4</sub> generated by methanogenic archaea is correlated to factors including glucose concentration, microbe density, pH (Bergman et al. 1998), plant species presence (Sutton-Grier and Megonigal 2011), and hydrology (Sha et al. 2011; Van der Nat and Middleburg 2000). Sediment depth, temperature, hydrology, and substrate quality are the most influential factors for methanogenesis (Grünfeld and Brix 1999; Wachinger 2000; Tian et al. 2011). Due to the number of non-constant factors that influence methanogenesis, there is great spatial variability in CH<sub>4</sub> generation within wetlands. For this reason, temperature, hydrology, and sediment depth were kept constant in this experiment.

This study involved monitoring methanogenesis and CO<sub>2</sub> emission rates from four different macrophyte beds and correlating the rates to species type and sediment properties. Cores were extracted from each of the four macrophyte beds. Sediment coring was used instead of an in-field gas trap because variability in CH<sub>4</sub> generation within a sediment core is less than that of emissions in the field. This is because the cores isolate microbial methanogenesis and do not account for vegetative CH<sub>4</sub> and CO<sub>2</sub> venting (Van de Nat and Middleburg 2000). Even with cores, although reduced, there is still spatial variability. To address this variability the sediments were characterized for nitrate, dissolved oxygen, dissolved organic carbon, and total organic carbon content; all of which impact CO<sub>2</sub> and CH<sub>4</sub> emissions.

## MATERIALS AND METHODS

### *Quantifying methane emissions from sediment cores*

#### *Site and species description*

The sampling sites were located in the Hudson River's Tivoli Bays National Estuarine Research Reserve. The bay is divided into a northern and southern bay with a railroad dike running North-South separating the river and the reserve. The *T. natans* bed is found in South Tivoli bay at the mouth of the Sawkill River. South Tivoli Bay is subtidal and is almost always inundated with water (Findlay et al. 1990). The *T. natans* bed is extremely dense and a majority monospecific (Tall et al. 2011). The *V. americana*, *T. angustifolia*, and *P. australis* beds are all within North Tivoli bay. North Tivoli Bay is a predominately intertidal marsh with the most prevalent genera being *Typha*, *Phragmites*, and *Lythrum* (Findlay et al. 1990). The *T. angustifolia* and *P. australis* are both present in the marsh while the *V. americana* is found in the channel of the river west of the railroad.

Each of the four species of interest is found throughout the Hudson River. The most dominant species in the river itself are *T. natans* and *V. americana* while *Typha* and *Phragmites* represent the major cover type in the intertidal wetlands (Nieder et al. 2011)

#### *Sampling*

16x4.8cm sediment cores were extracted from each of the four sites. To eliminate any hydrologic variability and to facilitate sampling, cores were taken at low tide. From each site, a total of six cores were collected within a 5 meter radius. Six cores were extracted to ensure an even representation of the spatial variability in sediment organic carbon content and other factors.

Cores were stored in five gallon buckets and water saturation was maintained according to tidal changes. The *V. americana* and *T. natans* cores were stored in one bucket that was completely inundated with water, mimicking the environments they inhabited. Water was changed twice weekly. *P. australis* and *T. angustifolia* cores were stored in one bucket in which the water levels were lowered below the surface of the sediment for three days and then subsequently inundated with water for three days. These cycles were repeated throughout the length of the experiment.

### *Experimental setup*

A gas-collecting water recirculation system was used to collect gas from the sediment cores. A peristaltic pump pulled water from a side-arm Erlenmeyer flask into the top a PVC pipe sediment core. The headspace between the rubber stopper and the sediment of the PVC core (112cm<sup>3</sup>) was filled with water and completely void of atmosphere. Any CH<sub>4</sub> generated from the core was dissolved into the water and the super-saturated effluent was pumped out of the core back into the side-arm Erlenmeyer flask. The Erlenmeyer flask had 115mL of gaseous headspace filled with inert helium. Every 30-60 minutes, 10mL gas samples were drawn from a stopcock in the mouth of the Erlenmeyer flask and 20mL of helium gas was pumped through the stopcock. Neutral pressure was immediately restored by venting the stopcock. Any water that was discharged into the flask was then pumped back out to the core to again become saturated with CH<sub>4</sub>.

### *Analysis*

Gas was sampled from the recirculation system every 30-60 minutes during a six-eight hour time period. The gas was sampled with a syringe and injected directly into a Shimadzu 14-A TCD Gas Chromatograph to quantify CH<sub>4</sub> and carbon dioxide content. To better assess variability in CH<sub>4</sub> and carbon dioxide emission rates from the sediment cores, each of the cores were characterized for dissolved oxygen (DO) content based on depth, dissolved organic carbon (DOC), and total organic carbon (TOC).

Nitrate and dissolved organic carbon were sampled from pore water extracted from homogenized sediment. The top 5.0cm of the sediment was used in the homogenization and the pore water was separated from the sediment using a centrifuge. The supernatant was then tested for nitrate using a SUNA v1.1 and the DOC content was analyzed using a Shimadzu 5050. Finally for TOC analysis 13.0 mL of sediment was massed, desiccated and massed again. Finally the ash-free-dry-mass (AFDM) was taken after combustion in a muffle furnace. The dissolved oxygen content was determined using an oxygen microelectrode.

### *Data management*

Emissions data from some cores were discarded based on one consistently applied rule. After a certain amount of time (around 200 minutes), CH<sub>4</sub> levels began to decrease in concentration within the flask headspace, defying the logical increase in concentration over time. This was attributed to the dilution of CH<sub>4</sub> from the extraction of a sample and reestablishment of neutral pressure with helium. This created an emission rate curve unrepresentative of the actual methanogenesis rates. CO<sub>2</sub> was much less sensitive to the dilution due to higher concentrations of CO<sub>2</sub> compared to that of CH<sub>4</sub>.

### *Statistical analysis*

After the slopes of the CO<sub>2</sub> and CH<sub>4</sub> emission rates for each core were calculated using a linear regression, normality was determined for each variable using a Shapiro-Wilk Test. *Phragmites* and *Trapa* data from the CH<sub>4</sub> emissions were non-normal while *Typha* data were normal. Due to the non-normal data, a Kruskal-Wallis test was used to compare multiple variables. *Vallisneria* was not included in the Kruskal-Wallis test because its values were so low they were estimated. *Phragmites*, *Trapa*, and *Typha* CH<sub>4</sub> emissions were not statistically different (P-value 0.137).

For the CO<sub>2</sub> data, only the *Vallisneria* data were non-normal. The Kruskal-Wallis test revealed that there was statistical significance between CO<sub>2</sub> rates between the four macrophyte species (P-value 0.000956). Nitrate data were all normally distributed so a single factor ANOVA test was used (P-value of 0.00769). DOC data were also normally distributed so an ANOVA was used (P-value of 0.129). The TOC data from

the *Vallisneria* sediment were non-normal while the others were parametric so a Kruskal-Wallis test was again used (P-value 0.000384).

A linear regression comparing total organic matter content to CO<sub>2</sub> and CH<sub>4</sub> emissions was also used.

## RESULTS

### *Methane and Carbon Dioxide emissions*

Mean CH<sub>4</sub> release from sediment cores was 11.0±9.7 ppm/112cm<sup>2</sup>/min for *P. australis*, 3.8±4.4 ppm/112cm<sup>2</sup>/min for *T. natans*, 6.5±9.2 ppm/112cm<sup>2</sup>/min for *T. angustifolia*, and ~0.32ppm/112cm<sup>2</sup>/min for *V. americana*. CH<sub>4</sub> emissions from the *V. americana* sediment were below detection limits and no visible peak was provided by the gas chromatograph for estimating peak value. Values were estimated from *V. americana* #2, the only sediment core from the *V. americana* stand that produced any quantifiable levels of CH<sub>4</sub> and should be viewed as the upper bound on CH<sub>4</sub> emission.

Mean CO<sub>2</sub> release from the sediment cores was the greatest from *P. australis* 15.5±8.3 ppm/112cm<sup>2</sup>/min, 10.5±4.5 ppm/112cm<sup>2</sup>/min for *T. natans*, 10.5±2.9 ppm/112cm<sup>2</sup>/min for *T. angustifolia*, and 1.2±0.9 ppm/112cm<sup>2</sup>/min for *V. americana*.

### *Oxygen saturation at depth*

O<sub>2</sub> had disappeared by 0.5 – 1 cm in all cores indicating they were all predominantly anoxic with no difference among plant communities.

### *Sediment TOC content*

The *P. australis* sediment cores had the highest average organic matter content of 29.3±5.6%. *T. angustifolia* had 17.8±6.4% organic matter, *T. natans* had 9.1±1.5%, and *V. americana* had the lowest at 4.1±1.4%.

### *Dissolved organic carbon content and pore water nitrate*

The pore water nitrate content of *T. natans* was the highest at 0.99±0.21mg/L, *V. americana* was 0.70±0.33 mg/L, *P. australis* was 0.47±0.07 mg/L, and *T. angustifolia* was 0.33±0.17 mg/L. *V. americana* had a DOC pore water content at 12.18±1.59 ppm, *T. angustifolia* had 10.79±5.77 ppm, *P. australis* had 10.53±1.09 ppm, and *T. natans* had 7.38±0.53 ppm (the DOC values lacked statistically significant differences).

## DISCUSSION

### *Methane emissions*

CH<sub>4</sub> emissions from each of the plant bed cores had very high standard deviations and there was significant overlap in the level of methanogenesis between each of the sediment types (excluding that of *V. americana*). This overlap may be attributed to the many variations in sediment conditions. One significant sediment characteristic that would help differentiate CH<sub>4</sub> emissions is the accessibility of organic matter to methanogens (Medvedeff et al. 2015; Sutton-Grier and Megonigal 2011). However, after measuring total organic carbon content in each of the sediment cores and comparing it to CH<sub>4</sub> emissions, there was no correlation between the two. Despite the significant differences in total organic matter content between the sediment types, the widespread spatial variability of other sediment

parameters may have resulted in the overlap between the plant bed CH<sub>4</sub> emissions (Sha et al. 2011). Temperature and water table fluctuations are the principal factors that influence methanogenesis (Grünfeld and Brix 1999; Sha et al. 2011; Van der Nat and Middleburg 2000) but since temperature and hydrology were controlled in this experiment, other sediment characteristics most likely contributed to the variability in CH<sub>4</sub> emissions within each plant community. The other contributors would be ethanol content, variations in microbial communities, pore water organic carbon content, pore water nitrate (acting as the more efficient terminal electron acceptor to small organic carbons), and oxygen saturation in the sediment.

Dissolved oxygen availability would have also contributed to variations in CH<sub>4</sub> emissions but the oxygen content did not differ among sediment types. However, there were differences in nitrate availability between the cores, which may have exhibited a strong control on CH<sub>4</sub> emissions. When nitrate is used in place of organic carbon as the terminal electron acceptor, the microbes produce CO<sub>2</sub> instead of CH<sub>4</sub>. The pore water nitrate content of the *P. australis* and *T. angustifolia* were the lowest of the four vegetation types which may have contributed to the higher mean CH<sub>4</sub> emissions. The *V. americana* had the second highest nitrate content at 0.70±0.33 mg/L which may be a contributing factor to the extremely low CH<sub>4</sub> emissions. As for *T. natans* the nitrate content was the highest; however, this may have been balanced out by the coupled high organic carbon content, resulting in overall CH<sub>4</sub> emissions greater than that of *V. americana* but lower than *P. australis* and *T. angustifolia* beds.

Although CH<sub>4</sub> levels could not be precisely estimated from *V. americana*, the sediment cores clearly yielded lower emission rates than the other macrophyte species. This may have been attributed to the relatively high nitrate content or the sediment morphology and location of this particular *V. americana* population. Because the stand was located in the middle of a Hudson River channel, much of the dead organic matter may have been immediately washed downstream instead of permeating the sediment for microbial use. Another explanation for the reduced *V. americana* CH<sub>4</sub> emissions is their root depth. Deeper roots undergo greater fermentation, thus expelling greater amounts of ethanol. When ethanol is present, it is bio-transformed into acetate. Since a majority of methanogens are acetoclastic, the release of CH<sub>4</sub> may be increased with more complex and deep root structures (Williams and Yavitt 2010). Roots of the other three species were deeper thus possibly contributing to greater acetate formation.

#### *Carbon dioxide emissions*

Unlike CH<sub>4</sub>, CO<sub>2</sub> emissions did show a positive correlation with increasing TOC levels, but that is because there is sufficient reason to believe that CO<sub>2</sub> emissions are less sensitive to spatial variability in sediment characteristics than that of CH<sub>4</sub>. *P. australis* emitted the greatest amount of CO<sub>2</sub> followed by *T. natans*, *T. angustifolia*, and *V. americana*. From this and other experiments it is reasonable to conclude that CO<sub>2</sub> emissions are greater than that of CH<sub>4</sub> from wetlands (Clair et al. 2002).

#### *Further considerations*

One reason why CH<sub>4</sub> production is limited and CO<sub>2</sub> is more prominent is that some CH<sub>4</sub> is consumed by methanotrophic archaea and bacteria (Chowdhury and Dick 2013 and Kao-Kniffin et al. 2010). Methanotrophy is present in the uppermost layers of wetland sediment and in the margins between sediment and water (Chowdhury and Dick 2013). The methanotrophs consume CH<sub>4</sub> and produce carbon dioxide as a byproduct. It is estimated that the aerobic oxidation of CH<sub>4</sub> mitigates 40-70% of global CH<sub>4</sub> emissions (Bridgham, et al. 2013) but this mainly accounts for terrestrial methanotrophic microbes (Chowdhury and Dick 2013). There is little to no research on the influence of wetlands as CH<sub>4</sub> sinks, mostly because wetland sediments are predominately anaerobic. Despite the anaerobic conditions, Chowdhury and Dick suspect that oxidation of CH<sub>4</sub> by methanotrophs is still significant but only vastly understudied (2013).

Another important consideration concerning methanogenesis is that CH<sub>4</sub> may also be transported through macrophyte tissue and vented out the leaves (Kao-Kniffin et al. 2010; Van der Nat and Middleburg 2000). Carbon dioxide is also vented out the leaves through plant respiration (Clair et al. 2002). In vegetated sediments, this form of plant-mediated transport, as opposed to ebullition, accounts for 62% of the CH<sub>4</sub> emissions (Grünfeld and Brix 1999). Despite the prominence of this pathway, only direct release of the two greenhouse gases from microbes to the water and atmosphere were accounted for in this study as live plants were not present in the cores.

Despite the variability and lack of significant difference in CH<sub>4</sub> emissions from each of the sediment types, it is still reasonable to conclude that certain macrophyte beds contribute to greater CH<sub>4</sub> emissions. For example, *P. australis* and *T. angustifolia* sediments could emit up to 50% and 30% respectively more CH<sub>4</sub> than *T. natans*. If recommendations were to be made for implantation of macrophyte species in artificial wetlands or to selectively preserve certain macrophyte species to minimize CH<sub>4</sub> emissions, it is reasonable to choose *V. americana* and *T. natans* as ideal species. They emit substantially less CH<sub>4</sub> than that of *P. australis* and *T. angustifolia*. As for CO<sub>2</sub> emissions, *V. americana* beds emit a significantly lower amount of CO<sub>2</sub> than any of the other species that were examined.

## CONCLUSION

Monitoring CH<sub>4</sub> flux from wetlands is an important but a difficult subject to study. Due to the influence that wetlands have on the global greenhouse gas budget, understanding to what degree certain wetland characteristics contribute to the greenhouse gas flux is imperative. Yet the spatial variability in wetlands leading to differential CH<sub>4</sub> emissions, even within a single plant population, makes studying greenhouse gas emissions challenging. There could be differences in sediment quality (amount of organic matter present, TOC and DOC), oxidation by methanotrophs, availability of alternative terminal electron acceptors, ethanol content, temperature, hydrology, pH, salinity, and densities and composition of microbe populations. Future studies should consider each and all of these factors when comparing sediment type to CH<sub>4</sub> emissions. Furthermore, this study only focused on the ebullition of CH<sub>4</sub> from sediments because it was conducted in a laboratory setting. Although there may be even greater environmental spatial variability because temperature, hydrology, and sediment depth are not controlled, gas traps may provide a better idea of CH<sub>4</sub> emissions from wetlands. This is in part due to the fact that up to 62% of CH<sub>4</sub> emissions from wetlands are vented through the vascular tissue of plants instead of bubbling from the sediment itself (Grünfeld and Brix 1999).

Although it is possible that carbon sequestration may outweigh methanogenesis in wetlands (Clair et al. 2002 and Kao-Kniffin et al. 2010), carbon sequestration and methanotrophy may not fully alleviate the potential long-term consequences of changing wetland greenhouse gas dynamics. Wetland CH<sub>4</sub> and CO<sub>2</sub> production remains an important process to study and should be of continued interest to the scientific community due to the impending threats of climate change. Climate change could exacerbate wetland greenhouse gas emissions, creating a positive feedback loop. The presence of salt from sea level rise, hydrologic changes, alterations in macrophyte species type, and temperature changes could potentially amplify the present trends of wetland greenhouse gas release (Clair et al. 2002; Grünfeld and Brix 1999; Medvedeff et al. 2015; Sutton-Grier and Megonigal 2011). These factors could also alter the efficacy of wetlands to act as carbon sinks and upset the balances between methanotrophy and methanogenesis (Stefanik and Mitsch 2014). To better understand and protect the environment, continued research on these important and dynamic ecosystems is imperative.

## ACKNOWLEDGEMENTS

Thank you to my mentor scientist Dr. Stuart Findlay for his tenacious assistance and guiding wisdom. Further thanks to David Fischer and Lisa Martel for their dependable support on sample analysis. None of

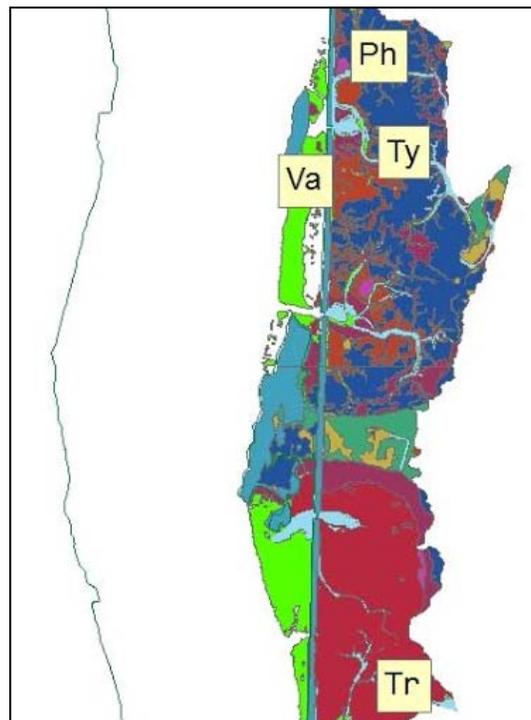
the research would have been possible if it weren't for the financial support of the National Science Foundation (NSF) and the support of the Cary Institute of Ecosystem Studies.

#### LITERATURE CITED

- Bergman, I., Svensson, B.H., and Nilsson, M. 1998. Regulation of methane production in a Swedish acid mire by pH, temperature and substrate. *Soil Biology and Biochemistry* **30**:729-741.
- Boon, P.I., Mitchell, A. 1995. Methanogenesis in the sediments of an Australian freshwater wetland: comparison with aerobic decay, and factors controlling methanogenesis. *FEMS Microbiology Ecology* **18**:175-190.
- Bridgham, S.D., Cadillo-Quiroz, H., Keller, J.K., and Zhuang, Q. 2013. Methane emissions from wetlands: biogeochemical, microbial, and modeling perspectives from local to global scales. *Global change biology* **19**:1325-1346.
- Chimney, M.J., and Pietro, K.C. 2006. Decomposition of macrophyte litter in a subtropical constructed wetland in south Florida (USA). *Ecological Engineering, The Everglades Nutrient Removal Project* **27**:301-321.
- Chowdhury, T.R., and Dick, R.P. 2013. Ecology of aerobic methanotrophs in controlling methane fluxes from wetlands. *Applied Soil Ecology* **65**:8-22.
- Clair, T.A., Arp, P., Moore, T.R., Dalva, M., and Meng, F.-R. 2002. Gaseous carbon dioxide and methane, as well as dissolved organic carbon losses from a small temperate wetland under a changing climate. *Environmental Pollution* **116**:143-148.
- Findlay, S., Howe, K., and Austin, H.K. 1990. Comparison of Detritus Dynamics in Two Tidal Freshwater Wetlands. *Ecology* **71**:288-295.
- Grünfeld, S., and Brix, H. 1999. Methanogenesis and methane emissions: effects of water table, substrate type and presence of *Phragmites australis*. *Aquatic Botany* **64**:63-75.
- Jespersen, D.N., K. Sorrell, B., and Brix, H. 1998. Growth and root oxygen release by *Typha latifolia* and its effects on sediment methanogenesis. *Aquatic Botany* **61**:165-180.
- Kao-Kniffin, J., Freyre, D.S., and Balsler, T.C. 2010. Methane dynamics across wetland plant species. *Aquatic Botany* **93**:107-113.
- Laure, T., Caraco, N., Maranger, R. 2011. Denitrification hot spots: dominant role of invasive macrophyte *Trapa natans* in removing nitrogen from a tidal river. *Ecological Applications* **21**:3104-3114.
- Levinton, J.S., and Waldman, J.R. 2006. *The Hudson River Estuary*. Cambridge University Press.
- Medvedeff, C.A., Inglett, K.S., and Inglett, P.W. 2015. Patterns and controls of anaerobic soil respiration and methanogenesis following extreme restoration of calcareous subtropical wetlands. *Geoderma* **245-246**:74-82.
- Ming, J., Xian-guo, L., Lin-shu, X., Li-juan C., and Shouzheng, T. 2007. Flood mitigation benefit of wetland soil - A case study in Momoge National Nature Reserve in China. *Ecological Economics* **61**:217-223.
- Mitsch, W.J., and Gosselink, J.G. 2000. The value of wetlands: importance of scale and landscape setting. *Ecological Economics* **35**:25-33.
- Miyajima, T., Wada, E., Hanba, Y.T., and Vijarnsorn, P. 1997. Anaerobic mineralization of indigenous organic matters and methanogenesis in tropical wetland soils. *Geochimica et Cosmochimica Acta* **61**:3739-3751.
- Nieder, W.C., Barnaba, E., Findlay, S.E.G., Hopkins, S., Holochuck, Nordica, and Blair, Elizabeth A. 2009. Distribution and abundance of submerged aquatic vegetation and *Trapa natans* in the Hudson River Estuary. *Journal of Coastal Research* **45**:150-161.
- Sha, C., Mitsch, W.J., Mander, Ü., Lu, J., Batson, J., Zhang, L., and He, W. 2011. Methane emissions from freshwater riverine wetlands. *Ecological Engineering, Special Issue: enhancing ecosystem services on the landscape with created, constructed and restored wetlands*. **37**:16-24.
- Sharifi, A., Kalin, L., Hantush, M.M., Isik, S., and Jordan, T.E. 2013. Carbon dynamics and export from flooded wetlands: a modeling approach. *Ecological Modelling* **263**:196-210.

- Sorrell, B.K., Downes, M.T., and Stanger, C.L. 2002. Methanotrophic bacteria and their activity on submerged aquatic macrophytes. *Aquatic Botany* **72**:107-119.
- Stefanik, K.C., and Mitsch, W.J. 2014. Metabolism and methane flux of dominant macrophyte communities in created riverine wetlands using open system flow through chambers. *Ecological Engineering, The Olentangy River Wetland Research Park: Two Decades of Research on Ecosystem Services* **72**:67-73.
- Sutton-Grier, A.E., and Megonigal, J.P. 2011. Plant species traits regulate methane production in freshwater wetland soils. *Soil Biology and Biochemistry* **43**:413-420.
- Tian, J., Chen, H., Wang, Y., and Zhou, X., 2011. Methane production in relation with temperature, substrate and soil depth in Zoige wetlands on Tibetan Plateau. *Acta Ecologica Sinica* **31**:121-125.
- Tong, C., Zhang, L., Wang, W., Gauci, V., Marrs, R., Liu, B., Jia, R., and Zeng, C. 2011. Contrasting nutrient stocks and litter decomposition in stands of native and invasive species in a sub-tropical estuarine marsh. *Environmental Research, Invasive Species* **111**:909-916.
- Van der Nat, F.-J., and Middelburg, J.J. 2000. Methane emission from tidal freshwater marshes. *Biogeochemistry* **49**:103-121.
- Wachinger, G., Fiedler, S., Zepp, K., Gattinger, A., Sommer, M., and Roth, K. 2000. Variability of soil methane production on the micro-scale: spatial association with hot spots of organic material and Archaeal populations. *Soil Biology and Biochemistry* **32**:1121-1130.
- Williams, C.J., and Yavitt, J.B. 2010. Temperate wetland methanogenesis: the importance of vegetation type and root ethanol production. *Soil Science Society of America Journal* **74**:317-325.

#### APPENDIX



**FIGURE 1.** Map of macrophyte coverage and sampling sites in the Tivoli Bays National Estuarine Reserve. The coordinates for the sampling beds are as follows:

Trapa: 589127,4651773

Typha: 589338,4655269

Phragmites: 589252,4655876

Vallisneria: 588855,4655173

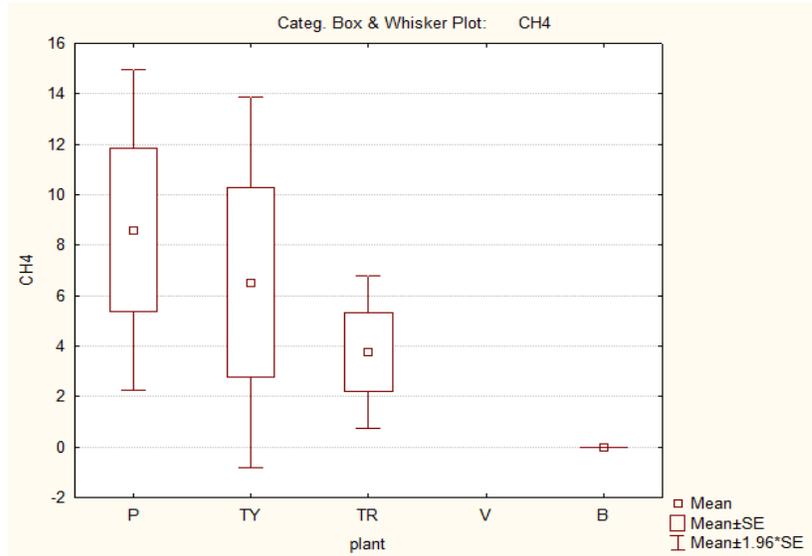


FIGURE 2. CH<sub>4</sub> emissions from each of the sediment types.

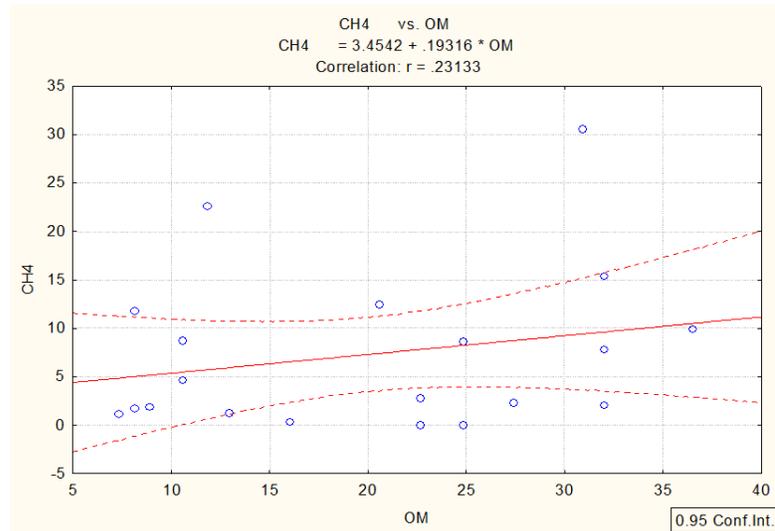


FIGURE 3. Comparing CH<sub>4</sub> emissions to organic matter content amongst cores.

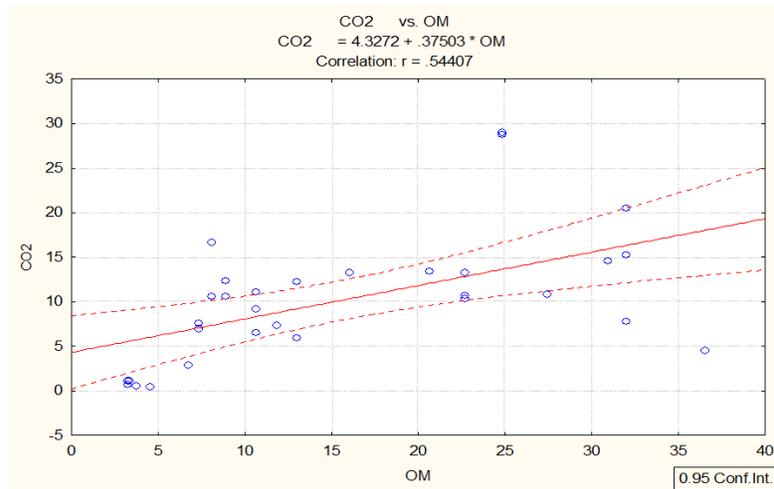


FIGURE 4. Comparing CO<sub>2</sub> emissions to organic matter content amongst cores.