

Name \_\_\_\_\_

Date \_\_\_\_\_

## **Questions for Nitrogen**

Read the article "Nitrogen" and answer the following questions:

- 1. What is the most common gas in the atmosphere? Why is this gas converted into other forms?
- 2. Why is nitrogen important? Give at least three examples of places where nitrogen exists.
- 3. Which nitrogen compounds are the most common in the environment?
- 4. What is the threshold of nitrate-nitrogen concentration for unpolluted water?
- 5. Give three examples of pollutants containing high levels of nitrate.
- 6. What is the consequence of excessive nitrate fertilization?
- 7. Where does the nitrate in the atmosphere come from?
- 8. What may happen when nitrate-nitrogen level in public water supplies exceeds 10mg/l?
- 9. Explain how high levels of turbidity in the Hudson River limit the threat of eutrophication.
- Imagine the molar concentration in a public water supply is 180µmol/l NO<sub>3</sub>. Convert it into a mass concentration and explain if whether or not we can drink it (according to the Environmental Protection Agency).